



# Air humidity calculation

<http://www.cactus2000.de>

## Constants:

$N_L$	$6.0221415 \cdot 10^{23}$	mol <sup>-1</sup>	Avogadro constant	NIST
$R$	8.31447215	J mol <sup>-1</sup> K <sup>-1</sup>	Universal gas constant	NIST
$M_{H_2O}$	18.01534	g mol <sup>-1</sup>	molar mass of water	
$M_{dry}$	28.9644	g mol <sup>-1</sup>	molar mass of dry air	

## Variables:

$c_{H_2O}$	g m <sup>-3</sup>	mass concentration of water
$[H_2O]$	cm <sup>-3</sup>	molecular concentration of water
$e$	hPa	vapor pressure of water
$P$	hPa	pressure
$P_{H_2O}$	hPa	partial pressure of water
$q$	kg kg <sup>-1</sup>	specific humidity (mass mixing ratio in wet air) <sup>1</sup>
$mmv$	kg kg <sup>-1</sup>	mass mixing ratio in dry air <sup>2</sup>
$n_{air}$	mol m <sup>-3</sup>	air density
$RH$	%	relative humidity
$T$	C	temperature
$T_D$	C	dew point temperature
$X_{H_2O}$	-	mole fraction, volume mixing ratio of water <sup>3</sup>

## Vapor pressure of water:

$$e = a_0 + T \cdot (a_1 + T \cdot (a_2 + T \cdot (a_3 + T \cdot (a_4 + T \cdot (a_5 + T \cdot a_6))))))$$

from Lowe, P.R. and J.M. Ficke, 1974: The computation of saturation vapor pressure. Tech. Paper No. 4-74, Environmental Prediction Research Facility, Naval Postgraduate School, Monterey, CA, 27 pp.

	<b>water</b>	<b>ice</b>
$a_0$	6.107799961	6.109177956
$a_1$	$4.436518521 \cdot 10^{-1}$	$5.034698970 \cdot 10^{-1}$
$a_2$	$1.428945805 \cdot 10^{-2}$	$1.886013408 \cdot 10^{-2}$
$a_3$	$2.650648471 \cdot 10^{-4}$	$4.176223716 \cdot 10^{-4}$
$a_4$	$3.031240396 \cdot 10^{-6}$	$5.824720280 \cdot 10^{-6}$
$a_5$	$2.034080948 \cdot 10^{-8}$	$4.838803174 \cdot 10^{-8}$
$a_6$	$6.136820929 \cdot 10^{-11}$	$1.838826904 \cdot 10^{-10}$

$$e = \min(e_{water}, e_{ice}), -50 \text{ C} \leq T \leq 100 \text{ C}$$

- 
- 1 multiply by 1000 to get 'g kg<sup>-1</sup>'
  - 2 multiply by 1000 to get 'g kg<sup>-1</sup>'
  - 3 multiply by 1000 to get 'per mille'

### **Other equations:**

$$\text{Air density: } n_{\text{air}} = \frac{P \cdot 100}{R \cdot (T + 273.15)}$$

$$\text{Relative humidity: } RH = \frac{e(T_D)}{e(T)} \cdot 100 = \frac{P_{\text{H}_2\text{O}}}{e(T)} \cdot 100$$

$$\text{Volume mixing ratio: } x_{\text{H}_2\text{O}} = \frac{P_{\text{H}_2\text{O}}}{P}$$

$$\text{Specific humidity: } q = \frac{x_{\text{H}_2\text{O}} \cdot M_{\text{H}_2\text{O}}}{x_{\text{H}_2\text{O}} \cdot M_{\text{H}_2\text{O}} + (1 - x_{\text{H}_2\text{O}}) \cdot M_{\text{dry}}}, \quad q = \frac{\text{mmv}}{1 + \text{mmv}}$$

$$\text{Mass mixing ratio: } \text{mmv} = \frac{q}{1 - q}$$

$$\text{Mass concentration: } c_{\text{H}_2\text{O}} = x_{\text{H}_2\text{O}} \cdot n_{\text{air}} \cdot M_{\text{H}_2\text{O}}$$

$$\text{Molecular concentration: } [\text{H}_2\text{O}] = x_{\text{H}_2\text{O}} \cdot n_{\text{air}} \cdot N_A \cdot 10^{-6}$$

### **Cactus2000:**

The equations on this sheet are used in the Cactus2000 'Air humidity converter':

<http://www.cactus2000.de/uk/unit/masshum.shtml> (in English)

<http://www.cactus2000.de/de/unit/masshum.shtml> (auf deutsch)

<http://www.cactus2000.de/fr/unit/masshum.shtml> (en français)